(Pages	:	4)	,
--------	---	----	---

Reg. No). :	 	•••	•••	٠,٠.	•••	• • •	••	• •	٠	••	•	٠.	
Name :		 												

Fifth Semester B.Sc. Degree Examination, February 2021.

First Degree Programme under CBCSS

Physics

Core Course VI

PY 1542 : QUANTUM MECHANICS

(2013, 2015-17 Admission)

Time: 3 Hours

Max. Marks: 80

SECTION - A

Answer any ten questions, each carries 1 mark

- 1. What is photo electric effect?
- Write de Broglie equation.
- Write the expression for energy of a harmonic oscillator.
- 4. Write the expression of Compton wavelength? Expression.
- In a harmonic oscillator write the expression for it ground state energy.
- 5. Why Rutherford's atom model is called planetary model?
- State any two properties of wave function.
- 8. State the uncertainty relation for energy and time.

- 9. Why ψ*ψ is always positive and real?
- 10. Define wave velocity.

 $(10 \times 1 = 10 \text{ Marks})$

SECTION - B

Answer any eight questions, each carries 2 marks

- 11. Explain Compton scattering.
- 12. Distinguish between phase velocity and group velocity.
- 13. What is normalization of a wave function?
- 14. What is a wave packet?
- 15. Write and explain the time dependent Schrodinger equation
- 16. Write the physical significance of $\psi(r,t)$ and it probability interpretation in short.
- 17. What are commuting operators?
- 18. What do you mean by expectation values?
- 19. Explain any two properties of Hermitian operators.
- 20. Explain Hamiltonian operators.
- 21. What is the condition in Hilbert space for which two functions are orthogonal?
- 22. What is momentum space wave function?

 $(8 \times 2 = 16 \text{ Marks})$

SECTION-C

Answer any six, each question carries 4 marks.

- 23. Ultraviolet light of wavelength 350 nm directed at a potassium surface. Find the maximum energy of photoelectrons. Given that the work function of potassium is 2.2 eV.
- 24. Find the de Broglie wavelengths of
 - (a) a 46-g golf ball with a velocity of 30 m/s, and
 - (b) an electron with a velocity of 107 m/s..
- 25. An electron is in a box of 0.1 nm, which is the order of magnitude of atomic dimensions. Find its permitted energies.
- 26. Show that the function $y = A \exp\left(-i\omega\left(t \frac{x}{v}\right)\right)$ is a solution of the wave equation
- 27. A particle limited to the x axis and has the wave function $\psi = ax$ between x = 0 and x = 1, all other regions $\psi = 0$.
 - (a) Find the probability that the particle can be found in the region 0.45 < x < 0.55
 - (b) Find the expectation value of $\langle x \rangle$ of the particles position
- 28. Show that the Eigen values of a Hermitian operator are real.
- 29. What is Hilbert space?
- Explain the properties of incompatible observables in quantum mechanics, give example
- 31. Briefly explain the finite square well scattering matrix.

 $(6 \times 4 = 24 \text{ Marks})$

SECTION-D

Answer any two, each question carries 15 marks

- 32. Discuss Bohr atom model. Derive the expression for energy levels and explain hydrogen spectrum.
- 33. Obtain the Time dependent and time independent form of Schrodinger equation.
- 34. Obtain the energy eigen values of a harmonic oscillator. Find the wave functions and the initial energy levels.
- 35. Explain the generalised statistical interpretation of quantum mechanical formulation.

 $(2 \times 15 = 30 \text{ Marks})$