H-1483

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Reg. No. :		

Third Semester B.Sc Degree Examination, October 2019 First Degree Programme under CBCSS

Physics

Core course

PY 1341-THERMODYNAMICS AND STATISTICAL PHYSICS (2013-2017 Admissions)

Time: 3 Hours

Name :

Max. Marks: 80

SECTION A

Answer all questions in one or two sentences. Each carries one mark.

- 1. If mean speed, most probable speed, r.m.s. speed of ideal gas molecules are 'a'. 'b' and 'c' respectively. Then correct relation is —————
 - (a) a=b=c
 - (b) a < b < c
 - (c) b<a<c
 - (d) b>a>c
- If dU = -dW in a thermodynamic process, then process is:
 - (a) Adiabatic
 - (b) Isothermal
 - (c) Isobaric
 - (d) Isochoric
- An example for Fermions is —————

- 4. What is the unit of thermal conductivity.
- 5. The efficiency of Carnot engine working between steam point and ice point is
- 6. What is the relation between thermodynamic probability Ω and entropy
- State Stefan's law.
- Define Helmholtz function.
- 9. State postulate of 'equal priori probabilities'
- Define phase space.

 $(10 \times 1 = 10 \text{ Marks})$

SECTION B

Answer any eight of the following. Each carries 2 marks. The answer should not exceed one paragraph.

- 11. State wiedemann Franz law.
- 12. State Liouville's theorem
- 13. If a drop of water falls on a very hot iron, it takes long time to evaporate. Why?
- 14. Distinguish first and second order phase transition?
- Write clausius clopeyron equation and explain the terms.
- Sketch Fermi-Dirac distribution function at (a) T=0 K and (b) T>0K
- 17. Write a note on ensembles.
- 18. What is Bose-einstein condensation?
- State and explain second law of thermodynamics.
- 20. Explain why the temperature of a gas drops in adiabatic expansion
- 21. How is the thermal conductivity of rubber determined by radial heat flow method?
- 22. What is the principle of increase of entropy?

 $(8 \times 2 = 16 \text{ Marks})$

SECTION C

.Answer any six of the following. Each carries 4 marks.

- 23. Define solar constant. How do you find the temperature of the sun from a knowledge of solar constant? (Radius of sun = 6..928x10⁵ km and R is mean distance of sun from the earth=148.48x10⁷ km).
- 24. A motor car tyre has a pressure of 3 atm at room temperature of 27°C. If tyre suddenly burst. What is the resulting temperature. $\gamma = 1.4$.
- 25. Show that work done along any adiabatic between two isothermals is independent of particular adiabatic.
- 26. A carnot engine working between source at 600 K and sink at T²K has an efficiency of 50%. If the temperature of source and sink are increased by 100 K, what is the efficiency of engine?
- 27. Obtain first and second Maxwell's thermodynamic equations.
- 28. Surface area of an electric heater is 350 cm². when heated its temperature becomes 727°C .How much energy will it radiate in one minute? Black body radiation may be assumed. Stefan's constant $\sigma = 5.67 \times 10^{-8} \text{ J.m}^2 \text{S}^{-1} \text{K}^{-4}$.
- 29. A collection of three non interacting particles shares 3 units of energy. Each particle is restricted to having an integral number of units of energy.
 - (a) How many macrostates are there?
 - (b) How many microstates are there in eaon of the macrostates?
 - (c) What is the probability of finding one of the particles with 2 units of energy?
- 30. Compute the Fermi energy $E_{\rm F}$ for sodium on the assumption that each sodium atom contribute one free electron to the electron gas. The density of sodium is 971 kg/m 3 and its atomic mass is 23 u.
- 31. The Latent Heat of Fusion of Ice I is 3.34x10⁵J/Kg at 0°C and atmospheric pressure. If the change in specific volume on melting is -9.05x10⁻⁵ m³/kg then calculate change in melting temperature due to change of pressure.

 $(6 \times 4 = 24 \text{ Marks})$

SECTION D

Answer any two questions Each carries fifteen marks.

- 32. Describe with necessary theory the construction and working of a diesel engine. Explain its merit over Otto engine.
- 33. Discuss brifly the Maxwell-Boltzmann, Bose-Einstein and Fermi-Dirac Statistics. Compare the three statistics.
- 34. Define Entropy. What is its physical significance? Show that entropy remains constant in a reversible process while increases in irreversible process?
- 35. Define Coefficient of Thermal conductivity? Describe an experiment with necessary theory to find out thermal conductivity of a bad conductor.

 $(2 \times 15 = 30 \text{ Marks})$